

PHARMACOLOGY I

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Doses of drugs in anaesthesia

Drugs in anaesthesia are commonly expressed in grams, milligrams or micrograms which refer to their mass.

Abbreviations are often used:

| Mass | Abbreviation |
|-----------|---|
| kilogram | kg |
| gram | g |
| milligram | mg |
| microgram | mcg or the Greek symbol mu is used, μg |
| nanogram | ng |

Examples: 1g flucloxacillin, 500mg thiopentone, 600mcg atropine. The words milli-, micro- and nano- which appear in front of “gram” refer to how many multiples of 10 are present.

- It is possible to convert from grams to nanograms as follows:

$$1\text{g} = 1\,000\text{mg} = 1\,000\,000\text{mcg} = 1\,000\,000\,000\text{ng}$$

- To convert, for example, from micrograms to milligrams:

$$600\text{mcg atropine} = 0.6\text{mg atropine.}$$

Recommended or therapeutic drug doses

Recommended or therapeutic doses enable one to calculate the correct dose of drug for the patient undergoing anaesthesia. They are usually expressed as mg/kg or mcg/kg. Note that this may also be written as $\text{mg}\cdot\text{kg}^{-1}$ or $\text{mcg}\cdot\text{kg}^{-1}$ and are calculated as follows.

Example: the dose of atropine is 20mcg/kg.

- To calculate the correct dose of drug for a patient, multiply the drug dose by the patient’s weight. In a 20kg patient we would give:

$$20\text{mcg/kg} = 20\text{mcg/kg} \times 20\text{kg} = 400\text{mcg} = 0.4\text{mg}$$
- To calculate the correct dose of atracurium (0.5mg/kg) for a 70kg adult. $0.5\text{mg/kg} \times 70\text{kg} = 35\text{mg}$

Maximum doses

Maximum doses may refer to local anaesthetic drugs like lignocaine or bupivacaine and indicate the maximum dose of drug that may be given to the patient safely without causing toxicity. In the case of local anaesthetics cardiac arrhythmias or convulsions may result if the maximum dose is exceeded.

Other units for describing drugs

Most commonly we describe the amount of drug present by reference to the mass of drugs (see above). However drug preparations may also be described by how many particles they contain. This gives an idea of the amount of drug present rather than the mass of the drug.

By convention the amount of a substance is measured in moles (abbreviation: mol). A mole has been defined by the Systeme International as the quantity of a substance that contains the same number of particles as there are atoms in 12g of carbon-12. There are 6.022×10^{23} atoms present in 12g carbon-12 and this equals one mole.

This dose method is often used for substances such as potassium (K^+) or sodium (Na^+) in the form of millimole (1mole = 1000 millimole - often written as mmol). This method is useful as sodium or potassium are often prepared with chloride and when administered it is helpful to consider only the amount of Na^+ or K^+ that is given. Therefore it is usually described in mmol.

Example: a solution of normal saline contains 154mmol Na^+ and 154mmol Cl^- in each litre.

Drugs in concentrations

When a substance is dissolved in a liquid it forms a solution. The volume of a solution is expressed in litres or millilitres. 1litre = 1000mls.

The substance dissolved is known as the solute. The amount of solute in a solution is expressed as a concentration. The amount of solute may be described by its mass (grams or milligrams per litre) or by its amount (moles per litre or millimoles per litre).

If the solute has a known chemical formula eg. salt (NaCl), then it is preferable to use mol/l or mmol/l. If

the solute does not have a defined chemical composition (such as a protein), then mg/l or g/l is used.

Some solutions such as local anaesthetics and thiopentone that are used in anaesthesia on a daily basis are expressed as a percentage eg. lignocaine 2% and thiopentone 2.5%. When using drugs prepared in this way it is necessary to calculate the number of mg in ml of solution. This is easiest done by multiplying the percentage of the solution by 10:

- 2% lignocaine x 10 = 20mg/ml
- 0.5% bupivacaine x 10 = 5mg/ml
- 2.5% thiopentone x 10 = 25 mg/ml

The maths behind this calculation is as follows:

- A 2.5% solution means that there is 2.5g of thiopentone in 100ml:

$$\begin{aligned} & 2.5\text{g in } 100\text{ml} \\ & = 2500\text{mg in } 100\text{ml} \\ & = 25\text{mg in } 1 \text{ ml (} 25\text{mg/ml)} \end{aligned}$$

Some solutions such as adrenaline (epinephrine) may be expressed as 1:1000 or 1:10 000 or 1:100 000. This means that in a 1:1000 adrenaline ampoule there is one part adrenaline to 1000 parts solution. To work out how many milligrams of adrenaline are present:

- 1:1000 solution adrenaline
 - = 1g adrenaline in 1000ml solution
 - = 1000mg adrenaline per 1000ml solution
 - = 1mg per ml
- 1:10 000 solution adrenaline
 - = 1g adrenaline in 10 000ml solution
 - = 1000mg per 10 000ml
 - = 1mg per 10ml which can also be expressed as 100mcg per ml

Preparing local anaesthetics with adrenaline

Pre-mixed ampoules of lignocaine 1% and 2% with adrenaline are sometimes not available and it may be necessary to prepare these solutions locally. A 1:200 000 solution means that there is 1 part adrenaline to 200 000 parts of solution (lignocaine in this instance).

In order to produce a 1:200 000 adrenaline solution, add 0.1ml adrenaline 1:1000 to 20ml lignocaine. The method is as follows:

- Take 1ml of adrenaline 1:1000 - dilute to 10mls with saline.

Take 1 ml of this mix which is now 1ml of 1:10 000 adrenaline. Add 19mls lignocaine. Total solution is now 20mls and the original adrenaline has been diluted 200 times = 1:200 000 solution

Other units - milliequivalents (mEq)

Sometimes the term milliequivalent is used in textbooks, although millimoles is the more correct expression. When substances are dissolved in a liquid, they may develop a charge. For example, when salt (NaCl) is dissolved in water, it separates into its two components: Na⁺ and Cl⁻ (these are known as ions). Sodium ions have one positive charge and chloride ions have one negative charge. The word milliequivalent refers to how many ions are present. In this case the number of milliequivalents will be the same as the number of millimoles in solution. Eg. a plasma sodium concentration of 140mEq/l equals 140mmol/l. However, in a magnesium chloride solution (MgCl₂) the magnesium ion has two positive charges so there will be two milliequivalents of magnesium per litre of solution. In this case the number of milliequivalents will not equal the number of millimoles.

Changes in drug names

New rules by the EEC now require the use of the Recommended International Non-proprietary Name (rINN) for drugs. In drugs where there is concern that a name change may pose a serious risk to patients, both the British Approved Name (BAN) and the rINN name will appear on the drug ampoule for at least the next 5 years. In other cases, the new name will appear alone.

Some examples of name changes that will affect anaesthetists are:

| UK name | New name (rINN) |
|----------------|-----------------|
| adrenaline | epinephrine |
| noradrenaline | norepinephrine |
| frusemide | furosemide |
| lignocaine | lidocaine |
| thiopentone | thiopental |
| phenobarbitone | phenobarbital |